



First synthesis of 3,3-difluoroserine and cysteine derivatives via Mg(0)-promoted selective C–F bond cleavage of trifluoromethylimines

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Abstract

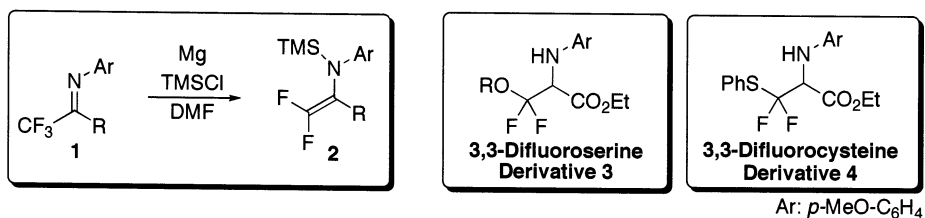
3,3-Difluoroserine and cysteine derivatives were synthesized via Mg(0)-promoted defluorination of trifluoromethylimines as a key step, followed by addition of alcohols and sulfenyl chloride, respectively. © 2000 Elsevier Science Ltd. All rights reserved.

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Among fluorinated amino acids, 3,3-difluoroamino acids are one of the most attractive synthetic targets, because some amino acids containing *gem*-difluoromethylene moiety have interesting biological activities¹ such as being an irreversible inhibitor of pyridoxal phosphate dependent enzymes.² In spite of the promising biological activities of 3,3-difluoroamino acids, only a few methods are available for the synthesis of these compounds.³ The development of a systematic synthesis of the target amino acids is required.

2-*N*-Silylamino-3,3-difluoroacrylate would be nucleophilically or electrophilically alkoxyated, thioalkoxyated and alkylated at the difluoromethylene carbon atom and thus would be a useful intermediate for the synthesis of various kinds of 3,3-difluoroamino acids. Meanwhile, as for the creation of difluoromethylene moiety from a trifluoromethyl moiety, we have already established electroreductive methods for *N*-silylated difluoroenamines **2**, using trifluoromethyl imines **1** as starting materials.⁴ Here, we describe the first synthesis of racemic 3,3-difluoroserine and 3,3-difluorocysteine derivatives **3** and **4** using a newly developed Mg(0)-promoted selective defluorination from trifluoromethyl imines **1** as a key step (Scheme 1). The Mg(0)-promoted defluorination provides many advantages for a practical preparation of difluoromethylene compounds, such as a less expensive process and the selective C–F bond cleavage from the trifluoromethyl group.⁵

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Scheme 1.

As shown in Table 1, the reaction of a variety of aromatic imines with either electron-withdrawing or electron-donating substituents on the aromatic ring (**1a–e**) and heteroaromatic imines (**1f**, **1g**) provided the desired enamines in excellent yields (entries 1–7). A ten gram-scale reaction proceeded in DMF at 0°C within 30 min using a commercially available magnesium^{6a} and no special activation of the magnesium such as iodine and dibromoethane or ultrasound treatment was needed. Meanwhile, the less reducible aliphatic imine required the higher temperature (70°C, entry 8). Moreover, the simplest 3,3-difluoroenamine **2i** (R=H) could be prepared in a reasonable yield (entry 9) and would be employed as a key starting substrate for fluorinated polyvinylamines. It is noteworthy that reductive defluorination of ethyl 2-(*N-p*-anisyl)imino-3,3,3-trifluoropropanoate (**1j**) gave ethyl 2-amino-3,3-difluoroacrylate (**2j**) in 50% yield in the presence of NaI and Et₃N (entry 10).^{6b}

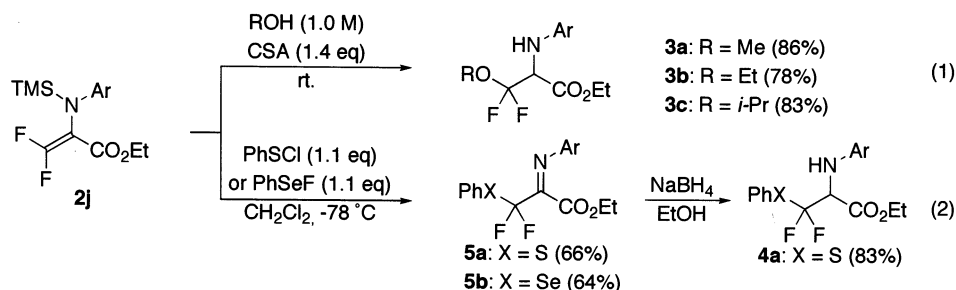
Table 1
Mg(0)-promoted reductive defluorination of trifluoromethylimines **1**^a

Entry	Imine 1	R	% Yield of 2 ^b	Entry	Imine 1	R	% Yield of 2 ^b
1	1a		82	6	1f		72
2	1b		88	7	1g		63
3	1c		quant.	8 ^d	1h	<i>n</i> -Hex	77 ^c
4	1d		quant.	9	1i	—H	47
5	1e		quant.	10 ^e	1j	—CO ₂ Et	50

^a The reaction was carried out by the procedure shown in note 6. ^b Isolated yield. ^c NMR yield, which was calculated by ¹⁹F NMR integration of product **2h** relative to 4,4'-difluorodiphenylmethane as an internal standard. ^d Conditions: 70 °C, 10 min. ^e 2.4 equiv of NaI and 4.2 equiv of Et₃N were used.

The fluorinated acrylate **2j** can be transformed into serine **3** and cysteine derivatives **4** as shown in Scheme 2. It is particularly noteworthy that **2j** reacted with both electrophiles and

nucleophiles smoothly without accompanying defluorination.^{7,8} The difluoromethylene carbon of **2j** is highly reactive even with a weak nucleophile like alcohols, meanwhile it can accept an electrophile such as PhSeF and PhSeCl at the β -carbon of the enamine **2j**. Thus, regioselective desilylative addition of various kinds of alcohols to **2j** proceeded smoothly under acidic conditions to give the corresponding 3,3-difluoroserine derivatives **3** in good yields (Eq. 1). **2j** was allowed to react with electrophiles such as benzenesulfonyl chloride or benzeneselenenyl fluoride,⁹ affording 3-thio- and 3-seleno-3,3-difluoro-2-iminopropanoates **5**. Reduction of thioester **5a** with NaBH₄ provided 3,3-difluorocysteine derivative **4a** in a good yield (Eq. 2).



Scheme 2. Synthesis of 3,3-difluoroserine and cysteine derivatives

It has been clarified that the enaminoester **2j** can react with both electrophiles and nucleophiles under conditions where no defluorination occurs from the difluoromethylene group. This result suggests **2j** is a promising precursor for a variety of difluoroamino acid derivatives.

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6. A typical reaction procedure is as follows; (a) TMSCl (18.2 mL, 143 mmol) and a solution of 2-(*N-p*-anisyl)imino-2-phenyl-1,1,1-trifluoroethane **1a** (10.0 g, 35.8 mmol) in freshly dried DMF (43 mL) were added to a suspension of magnesium (6.97 g, 287 mmol) in DMF (100 mL) at 0°C under an argon atmosphere. After stirring for 30 min at 0°C, TMSCl was removed under reduced pressure (22 mmHg) at rt and a 1/9 solution of Et₃N/hexane (100 mL) was added. Magnesium was separated by filtration through Celite[®] under reduced pressure and the filtrate was washed with H₂O (200 mL). The aqueous layer was extracted with hexane (100 mL×2) and the organic layer was washed with H₂O (50 mL) and then dried over MgSO₄. Evaporation of the solvent and bulb-to-bulb distillation (0.3 mmHg, 120°C) provided **2a** in 82% yield. (b) TMSCl (40.0 mmol) and a solution of ethyl 2-(*N-p*-anisyl)imino-3,3,3-trifluoropropanoate **1j** (10.0 mmol) in freshly dried DMF (10 mL) were added to a suspension of magnesium (80.0 mmol), Et₃N (42.0 mmol) and NaI (24.0 mmol) in DMF (30.0 mL) at 0°C under an argon atmosphere. After stirring for 30 min at 0°C, magnesium was separated by filtration through Celite[®] under reduced pressure and the filtrate was extracted with hexane (20 mL×3). Evaporation of solvent and bulb-to-bulb distillation (0.25 mmHg, 90°C) provided **2j** in 50% yield.
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